Ti(III)-MEDIATED REDUCTIONS OF N-HYDROXY IMIDAZOLES TO 1-PROTIO-2,4(5)-DISUBSTITUTED DERIVATIVES

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SUMMARY: N-Hydroxy imidazoles are converted via TiCl₃ in MeOH/H₂O to N-protio derivatives, useful precursors of chiral amino acids.

The unique role which appropriately functionalized imidazoles play as 'azabutadienes' in reactions with singlet molecular oxygen has led to their conversion to amino acid derivatives in optically active form.¹ This novel method, which equates an apparently simple heterocycle with either naturally occurring or synthetic amino acids, requires a 1.2.4-trisubstitution pattern on the ring. Although improvements on the well-established Weidenhagen route to 2,4(5)-disubstituted materials have been made,² we took note of a rather sketchy communication which surfaced over a decade ago describing the preparation of di- and trisubstituted imidazoles using commercially available nitrosonium tetrafluoroborate (NOBF₄) in nitrile media.³ This protocol is especially attractive as it relies on olefin precursors and involves only two operations. Unfortunately, in our hands, while the N-hydroxy imidazoles mentioned in this report were easily realized, their reduction to the necessary 1-protio species failed completely under literature 'conditions' (Red-Al, rt), as well as upon examination of numerous other reagents and conditions.⁴ Based on the



successful reduction of other N-hydroxy-containing compounds (e.g., pyrimidines,⁵ oximes,⁶ and hydroxamic acids⁷) with TiCl₃, it was hoped that imidazoles <u>1</u> would follow suit. Much



R	Yield (%)
PhCH ₂	83
t - Bu	94
(CH ₃) ₂ CHCH ₂	87
$\underline{n} - C_{a}H_{q}$	94
$\underline{n} - C_5 H_{11}$	92

to our delight, when a methanolic solution of $\frac{1}{2}$ (<u>ca</u>. 0.2 M) was treated with TiCl₃ (2 equiv, 20% in H₂O) at room temperature, TLC analysis indicated complete consumption of starting material. Subsequent quenching with saturated aqueous NaHCO₃ and extractive workup (CH₂Cl₂) afforded the desired products (83 - 94 %, isolated) after filtration through SiO₂ (10% EtOH/CHCl₃). Typical substrates are illustrated in Table I and demonstrate the generality of the method. In light of this highly efficient conversion of $\frac{1}{2}$ to 2,4(5)-disubstituted imidazoles, a straightforward two-step route to amino acid and dipeptide precursors 1,2 is now in hand.

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References and Notes

- 1. Lipshutz, B.H., Morey, M.C., J. Am. Chem. Soc., in press.
- 2. Lipshutz, B.H., Morey, M.C., J. Org. Chem., <u>48</u>, 3745 (1983).
- 3. Scheinbaum, M.L., Dines, M.B., <u>Tetrahedron Letters</u>, 1971, 2205.
- 4. Some reagents included: Zn HC1, LAH, H2, Pd C, A1(Hg), Na(Hg), Zn/HOAc, BH3 THF.
- 5. McCall, J.M., TenBrink, R.E., Synthesis, 1975, 335.
- 6. Timms, G.H., Wildsmith, E., <u>Tetrahedron</u> Letters, 1971, 195.
- 7. Mattingly, P.G., Miller, M.J., J. Org. Chem., 1980, 45, 410.

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